

ENHANCED PROPERTIES FROM MIXING NATURAL RUBBER WITH RECYCLED POLYVINYL CHLORIDE BY MELT BLENDING

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ABSTRACT

The growing consumption of polymer products has created large quantities of waste materials resulting in environmental issues. Researchers have started to focus on recycling materials, not only for environmental reasons but also to decrease the cost. Polyvinyl chloride (PVC) is a universal non-biodegradable polymer which can be processed into a variety of long-life and short-life products. Blending with biodegradable natural rubber is a nice combination to convert composites into biodegradable products. In this study, natural rubber/recycled PVC composites, with different percentage loadings of each material, were produced using a two-roll mill to ascertain the optimum properties of the blends by analyzing the mechanical and thermal properties. The hardness and the modulus were enhanced with the increasing r-PVC loading. The elongation at break increased with the r-PVC loading but decreased dramatically after 40% r-PVC loading while the tensile strength decreased after the 60% r-PVC loading. In the thermal analysis, the crystallization temperature and melting point increased with the r-PVC loading.

Keywords: Natural rubber; recycled PVC; mechanical properties; thermal properties.

INTRODUCTION

Natural rubber (NR) is an elastomer produced from the latex of the rubber tree. NR, as a natural renewable resource, has many excellent comprehensive properties, such as outstanding resilience, high strength and good processability [1-3]. However, as an unsaturated polymer, NR tends to gradually degrade at higher temperatures or when exposed to oxygen, ozone or ultraviolet, which has a significant negative effect on its special applications [4-7]. Its low mechanical properties, with low surface energy and low barrier properties, are one of the reasons NR must be blended with other materials to enhance the properties [8-11]. Unfortunately, there is a lack of research focusing on blending NR with recycled materials. A review by Huang et al. [12] into the application of solid waste material (SWM) in the UK for asphalt pavements provides an overview on recycling materials. They covered the waste from glass, steel slag, scrap tires and plastics. Tire rubber is used in asphalt mixtures to reduce cracking, improve durability and mitigate noise. Recycled plastics can either replace a portion of the aggregates or serve as a binder modifier, such as for low density polyethylene (LDPE) which is replacing 30% of the 2.36–5 mm aggregates, reduces the mix density by 16%, shows a 250% increase in Marshall Stability and improves the indirect tensile strength (ITS) in ‘Plastiphalt’ mixtures [13]. Recycled 0.30–0.92-mm size LDPE, replacing 15% of the

aggregates in asphalt surfacing, has increased the stability retained (SR) by 15%, and enhanced the rutting and water resistance. Recycled PVC bottles contribute to increasing the bending strength by adding 2–6% of the mixture by weight. Recycled polyethylene (PE) film is used for 0.4% of the mixture by weight (about 8% of the binder weight) as a bitumen modifier, increasing the Marshall Stability before and after logging.

PVC is one of the most commonly used thermoplastic materials in worldwide polymers. Because of the particular properties of PVC, such as its low cost and high performance, and the high possibility of producing a variety of products from different processing conditions and techniques, PVC has become the polymer of choice [14-18]. Recently, the question of the disposal of PVC waste resulting from its rapid growth has resulted in increased public awareness. One of the simple routes for disposing of PVC waste is landfill but it has become more expensive and, in many countries, this method of disposal is no longer approved due to increasing consumption, the lack of landfill areas, and the environmental hazards associated with the chlorine content of the polymer [19-22]. A more suitable route without environmental problems, and in combination with recovering the polymer energy and material content, is recycling. Until now, researchers and industrialists have been funded in this field. They have proposed that PVC can be successfully recycled into several products, such as bottles, pipes, and pipe fittings among others, with good appearance and properties. PVC is now one of the largest recycled polymers by volume in developed countries as it has attracted attention in the fields of research and technology.

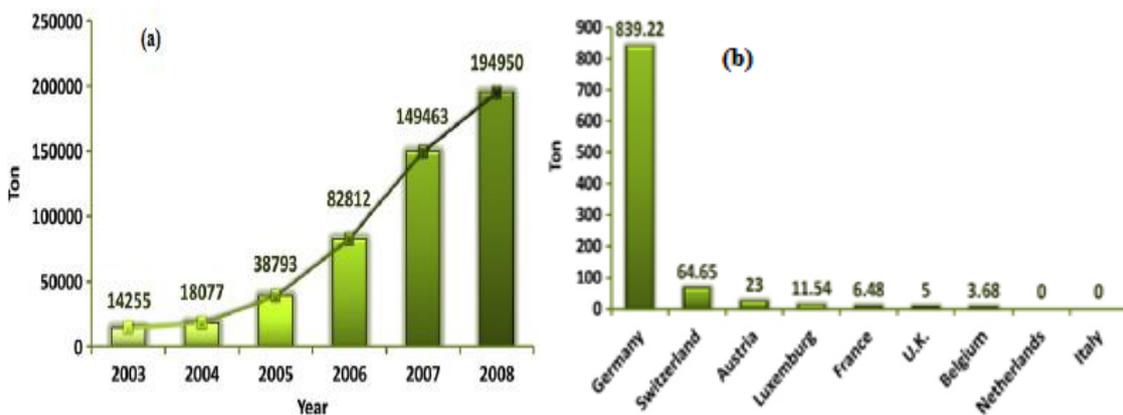


Figure 1. (a) The tonnage of recycled PVC in Europe from 2003 until 2008; (b) The tonnage of roofing materials recycled in Europe by country in 2008 [16].

Although recycling has taken place, it should be stressed here that many plastics cannot simply be separated or processed. Moreover, the mixed recycled material would have poor mechanical properties compared to the original material and only a narrow range of possible applications. Currently, the recycling operations for polymers, PVC in particular, are increasing, especially in the following aspects: development of techniques and instrumentation for the separation of PVC from waste streams, improvement of the current methods or development of new methods, research into the compatibility of recycled PVC with other polymers and virgin PVC, recycling of mixed PVC waste and enhancement of the physical and mechanical properties of recycled PVC [16]. This study aims to justify the mechanical properties of unfilled NR and

NR/recycled PVC (NR/r-PVC) blends. The thermal properties of all samples will also be analyzed. The main purpose of this study is to reduce PVC waste and convert it into more valuable materials. The second rationale is due to the fire resistant behavior of PVC after that of polytetrafluoroethylene (PTFE) [23]. This superiority meets the shortage of NR which is easily burned. The oxygen index is used to evaluate the fire retardant properties of plastic. Table 1 shows the comparison of the oxygen index between PVC and other polymers: PTFE, polycarbonate [24], polyethylene terephthalate (PET), polystyrene (PS), polypropylene (PP) and polyethylene (PE). Since the oxygen concentration in the air is 21%, a plastic with an oxygen index greater than 22 is classified as a self-extinguishing property, while a plastic with an oxygen index of less than 21 is flammable.

Table 1. The oxygen index of various plastics [23].

Materials	PTFE	PVC	PC	PET	PS	PP PE
Oxygen Index	95.0	45~49	26~28	20.0	17.6~18.3	17.4 17.4

Experimental Procedure

Standard Malaysia Rubber 10 (SMR-10) NR was supplied by Lee Rubber Sdn. Bhd., Kelantan and recycled PVC was obtained from PVC matting. Other materials were obtained commercially. Three major steps were involved: preparation of the samples, mixing and characterization.

Preparation of Samples

The recycled PVC (r-PVC), obtained from recycled PVC matting, was washed with distilled water and dried in an oven overnight. Then, the material was blended using an IKA basic analytical mill in order to grind it. The blended r-PVC was sieved using a 112- μ m sized sieve.

Mixing Process

Five samples were prepared with different NR and recycled PVC loadings (by weight percentage, wt.%): 100 NR, 80/20 NR/r-PVC, 60/40 NR/r-PVC, 40/60 NR/r-PVC and 20/80 NR/r-PVC as shown in Table 2. The samples were compounded via melt blending using a two-roll mill at 35–40 °C and four chemicals of fixed amounts were added continuously during mixing to assist in acceleration and crosslinking. Initially, the NR was masticated on the mill for 15 min. Thereafter, the rubber was blended with zinc oxide, stearic acid, CBS, r-PVC and sulfur. The mixture was plasticized for 10–15 min and thin-passed several times at 70–80 °C. For higher r-PVC loadings (60% and above), the r-PVC became a matrix and the NR was a filler. First, the r-PVC was melted with a melting point of around 160 °C, followed by the addition of zinc oxide, stearic acid, CBS, NR and sulfur. At this stage, the temperature was decreased to around 70–80 °C. Last, the obtained samples were hot-pressed using a 1-mm thick mold to form sheet samples.

Table 2. The composition of the samples.

Sample, NR/r-PVC	Chemical compositions added to all samples (%)
100	Zinc oxide (5%); stearic acid (2%);
80/20	cyclohexylbenzothiazole-
60/40	sulfenamide (CBS) (2.5%); sulfur
40/60	(0.5%)
20/80	

Characterization Process

All five prepared samples were tested using Shore durometer hardness testing in accordance with ASTM D2240 [25] to measure the resilient material’s resistance to indentation. Each sample was tested three times for measurement accuracy. During the hardness testing, the sample was put on a flat surface. The Shore durometer hardness tester was pressed onto the sample. After 15 seconds, the reading was shown by durometer. Tensile testing measures the force required to break a plastic sample and the extent to which the sample stretches or elongates to that breaking point. The samples were cut into dumbbell shapes using the ASTM D412 standard cutter. The tensile test was undertaken using an Instron Universal Testing Machine. Thermal analysis was undertaken using differential scanning calorimetry (DSC) employing a Mettler Toledo to determine the temperature of a sample to glass transition (T_g), crystallization temperature (T_c) and melting point (T_m). During DSC characterization, the weighed sample was put in an aluminum pan clamped using the clamp pan in the DSC equipment and the experiment was run at temperatures in the range -80 °C to 500 °C at a heating rate 10 °C /min.

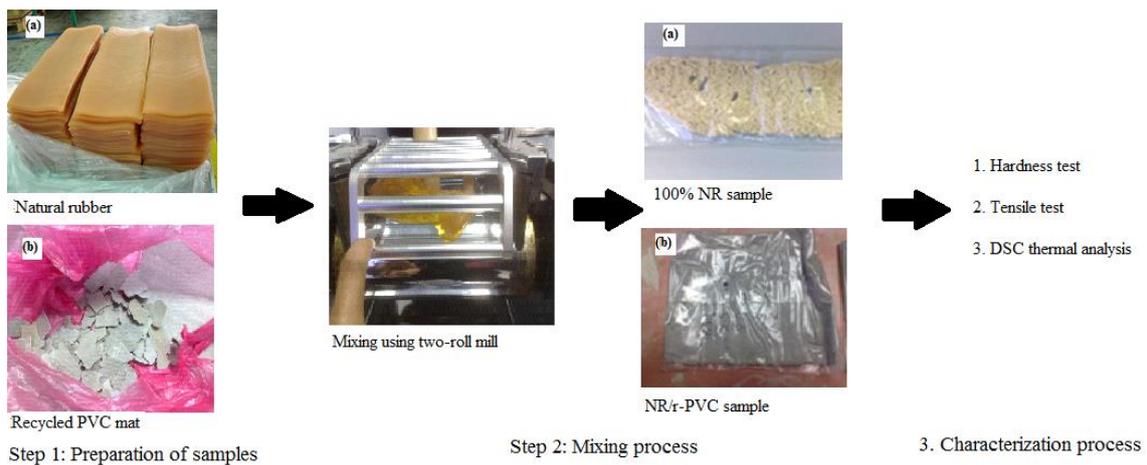


Figure 2. Schematic diagram of the preparation of the samples, mixing using a two-roll mill and characterization.

RESULTS AND DISCUSSION

Mechanical Properties of Samples

The mechanical properties of the unfilled NR and NR/r-PVC blends are presented in Table 3. The hardness and the modulus were enhanced with the increased r-PVC loading. The incorporation of r-PVC into the rubber matrix increased the stiffness of the vulcanizates, given that the tensile modulus represents the material stiffness [1]. The reinforcement effect of r-PVC, such as hardness, modulus and tensile strength, may be explained by the interaction between the rubber and r-PVC chain. The elongation at break was increased by the r-PVC loading but decreased dramatically after 40% r-PVC. The tensile strength also showed a decrease after 60% r-PVC loading due to the increased aggregation of r-PVC in the matrix. These results showed similar findings to those of other researchers [26]. The increase in tensile strength and elongation at break were caused by the filler particles being fully wetted by the NR matrix, especially at low loading. The rubber chain could penetrate into the pores of the r-PVC filler and become a part of the filler (bound rubber). The further addition of r-PVC as a filler decreased the tensile strength and elongation at break. This was caused by agglomeration in the filler particles, thereafter being less wetted and less penetrated by the NR chains, which had a detrimental effect on the tensile strength and elongation at break [27].

Table 3. Mechanical properties of samples.

Sample, NR/r-PVC (%)	Hardness (Shore A)	Elongation at break (mm)	Tensile strength (MPa)	Tensile modulus (MPa)
100	10.8	349	20	1.9
80/20	22.3	380	24	2.0
60/40	32.5	153	24.5	4.7
40/60	49.5	87.8	20	8.6
20/80	33.3	17.5	18.4	8.8

Thermal Analysis of Samples

DSC measurements were taken for all blends in order to study the influence of both nature and the content of r-PVC on the glass transition temperature of the NR matrix. Compared to the neat NR matrix, no significant variation in T_g on the r-PVC addition was reported regardless of the nature and content of the r-PVC [28, 29]. Recycled PVC itself is a polymer and contains plasticizer which is added during the manufacture of PVC matting. The crystallization temperature and melting point increased due to the r-PVC loading. At 60% r-PVC loading, the T_m increased to 416.20 °C, due to the PVC being highly fire resistant. At the 80% r-PVC loading, a slight decrease in T_c and T_m was shown. This can be explained by the aggregation causing the materials to be no longer compatible with each other. The viscosity of the blends increased because of the poor flowability of the neat NR; hence, the diffusion rate of the r-PVC chains was reduced [27].

Table 4. DSC analysis results for NR and its blends.

Sample, NR/r-PVC (%)	Glass transition, T_g (°C)	Crystallization temperature, T_c (°C)	Melting point, T_m (°C)
100	-60.98	198.08	377.08
80/20	-61.46	266.51	340.57
60/40	-62.33	270.51	360.17
40/60	-62.03	286.84	416.20
20/80	-63.80	272.40	333.35

CONCLUSIONS

Recycling of waste materials is compulsory as a result of global environmental issues caused by the increase in waste over the years. In this study, NR and NR/r-PVC blends were produced by melt compounding and compression molding. The effect of r-PVC loading on the mechanical and thermal properties was studied. This research showed results in parallel with other studies as noted in the discussion. The hardness reached its maximum at 40/60% NR/r-PVC blending, while the tensile strength showed a decreasing trend after 60% r-PVC loading. The tensile modulus was enhanced due to the r-PVC loading. The glass transition temperature showed no significant variation. Furthermore, both T_c and T_g gave the highest result for 40/60% NR/r-PVC blending, with a crystallization temperature of 286.84 °C, and a melting point temperature of 416.20 °C.

ACKNOWLEDGEMENTS

The authors gratefully acknowledge the Faculty of Chemical Engineering Technology, TATI University College for providing laboratory facilities and giving supports in this study.

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